# Critical Review and Screening of Laboratory Supplies for PFAS Analysis in Water Samples Landon Wiest, Shun-Hsin Liang, Mike Chang; Restek Corporation

## Introduction

The ubiquitous nature of PFAS in the environment makes ensuring a contaminant-free workflow (Figure 1) essential. In this presentation, we demonstrate that Resprep S-DVB SPE cartridges and related sample preparation products are consistently free of background interferences. In addition, a PFAS delay column effectively removes any contamination that may be present in the instrument. Using the materials and procedure presented here, EPA Method 537.1 requirements for cleanliness, accuracy, and precision were reliably met.

#### Figure 1: Sample preparation following EPA Method 537.1

Conditioning	<ul> <li>Mount Resprep S-DVB SPE cartridges (6 mL, 500 mg) onto the</li> <li>Pass 15 mL MeOH through the cartridges.</li> <li>Pass 18 mL reagent water through the cartridges.</li> <li>Mount the sample reservoirs onto the SPE cartridges as show in</li> </ul>
Extraction	<ul> <li>Load the samples into the reservoirs and establish a flow rate of After the samples pass through the cartridges, rinse the sample the sample reservoirs with 7.5 mL of reagent water.</li> <li>Rinse the sample bottles, and then the sample reservoirs with a aliquot of reagent water.</li> <li>Dry the SPE cartridges by drawing air through the manifold for \$ (10-15 in. Hg).</li> </ul>
Elution	<ul> <li>Place the collection tubes in the manifold and elute the analytes aliquots of MeOH.</li> <li>Remove the collection tubes from the manifold and concentrate dryness at 65°C under a gentle flow of nitrogen.</li> <li>Reconstitute with 1 mL 96:4 MeOH:water and add internal standmix.</li> <li>Transfer to polypropylene vials and cap with polyethylene caps.</li> </ul>

## Methods and Results

Laboratory reagent blanks were prepared for three different lots of Resprep S-DVB SPE cartridges

- $\succ$  Entire workflow, including manifolds, sample vials, solvents, reservoirs and connectors were free of PFAS contaminants tested in EPA Method 537.1 (Figure 2) via LC-MS/MS on a Shimadzu 8045.
- $\succ$  The use of a PFAS delay column provided excellent chromatography, by retaining any PFAS contaminants in the LC
- Raptor C18 analytical LC column gives excellent selectivity for PFAS compounds (Figure 3).
- Recoveries and %RSDs meet all acceptance criteria for EPA Method 537.1 (Table I)

Pure Chromatography



vacuum manifold.

in the figure 2

of 10-15 mL/min. bottles – and then

second 7.5 mL

5 min at high vacuum

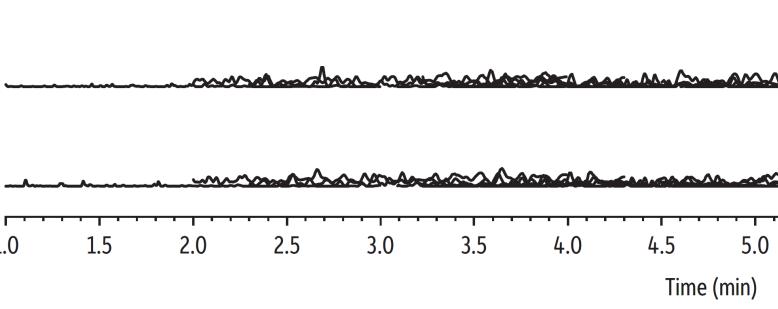
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the extracts to

dard; then, vortex to

#### Figure 2: Reagent Blanks (LRB) from the Entire Extraction Workflow with Multiple Lots of S-DVB SPE



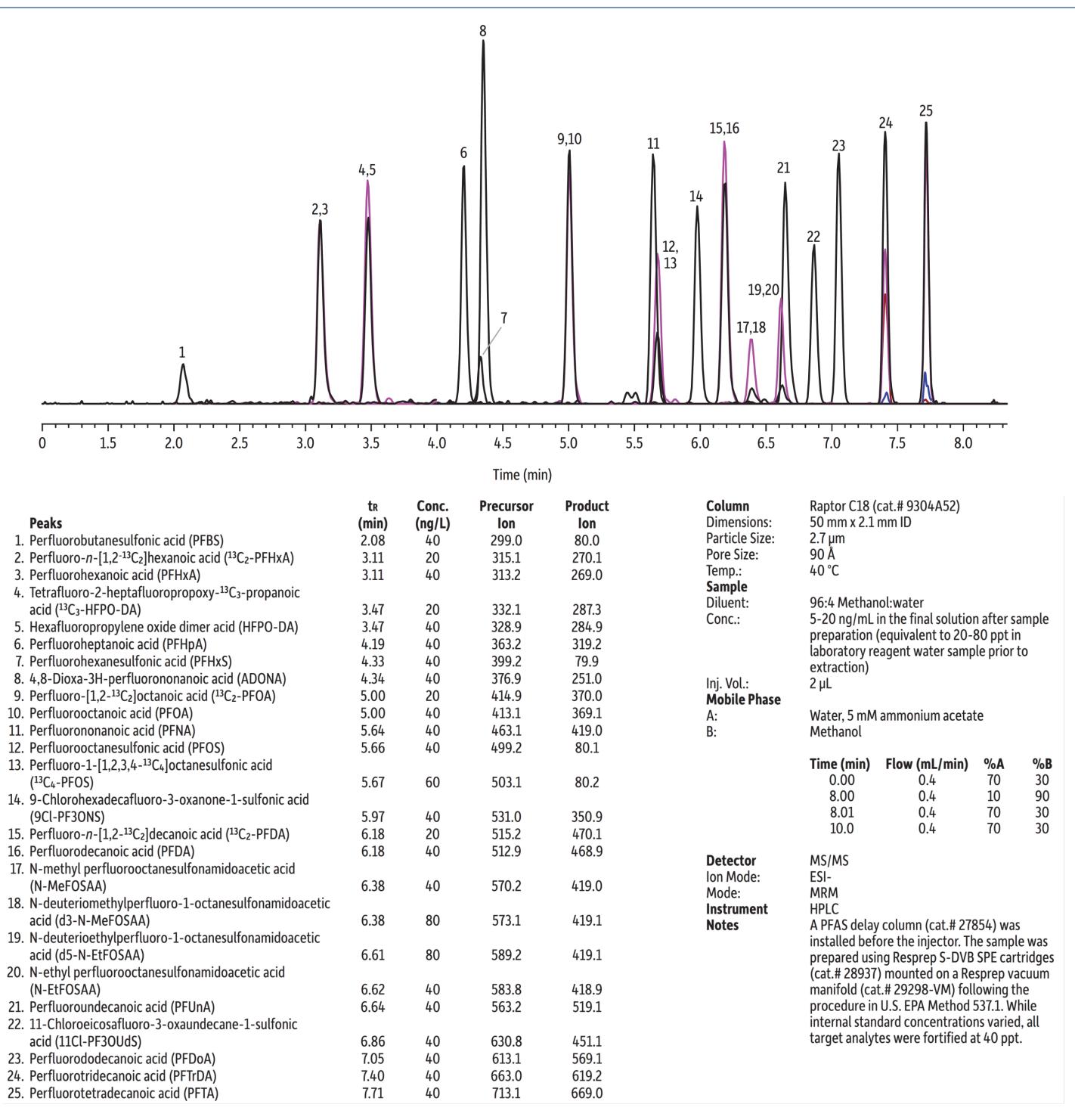


### Table I: Precision and Accuracy Results for Method 537.1 PFAS Analysis (n = 4)

Analyte	%RSD*	Mean Recovery**
Perfluorobutanesulfonic acid (PFBS)		91.1%
Perfluorohexanoic acid (PFHxA)		99.4%
Hexafluoropropylene oxide dimer acid (HFPO-DA)		94.4%
Perfluoroheptanoic acid (PFHpA)		92.7%
Perfluorohexanesulfonic acid (PFHxS)	11.9%	89.4%
4,8-Dioxa-3H-perfluorononanoic acid (ADONA)	5.18%	96.6%
Perfluorooctanoic acid (PFOA)	5.21%	91.6%
Perfluorononanoic acid (PFNA)		97.2%
Perfluorooctanesulfonic acid (PFOS)		87.8%
9-Chlorohexadecafluoro-3-oxanone-1-sulfonic acid (9CI-PF3ONS)		85.1%
Perfluorodecanoic acid (PFDA)		93.6%
N-methyl perfluorooctanesulfonamidoacetic acid (N-MeFOSAA)		82.8%
N-ethyl perfluorooctanesulfonamidoacetic acid (N-EtFOSAA)		106%
Perfluoroundecanoic acid (PFUnA)		97.5%
11-Chloroeicosafluoro-3-oxaundecane-1-sulfonic acid (11CI-PF3OUdS)		87.6%
Perfluorododecanoic acid (PFDoA)		99.0%
Perfluorotridecanoic acid (PFTrDA)		89.1%
Perfluorotetradecanoic acid (PFTA)		89.7%

B 2PE		
	Description	Restek Cat. #
	Resprep S-DVB SPE (6 mL, 500 mg)	28937
	Resprep Vacuum Manifold (12 or 24 port)	) 28298-VM, 28299-VM
	Reservoirs (Polypropylene)	26015
	Connectors (Polypropylene)	26007
	Vials (Polypropylene)	23245
	Vial Caps (Polyethylene)	23247
		Lot Lot
2.5	<b>3.0</b> 3.5 4.0 4.5 5.0 5.5 6.0	Lot 6.5 7.0 7.5 8.0
	Time (min)	

#### Figure 3: Laboratory Fortified Blank (LFB) at Midrange (40ppt)



## Conclusions

- Eliminating PFAS contaminants from workflows is difficult
- method that is free of PFAS contaminants.
- Method.

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registered in other countries.

 $\succ$  The above conditions and products provide an entire sample preparation workflow and HPLC

> The entire workflow meets the criteria for drinking water analysis as set forth by the EPA 537.1